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POGIL Chemistry Teachers Edition

$\text{Mg(NO}_3)_2(\text{aq}) + 2\text{H}_2\text{O}(\text{l}) \rightarrow \text{MgCO}_3(\text{s}) + 2\text{NaCl}(\text{aq})$
 $\text{FeS}(\text{s}) + 2\text{HCl}(\text{aq}) \rightarrow \text{H}_2\text{S}(\text{g}) + \text{FeCl}_2(\text{aq})$
 $\text{HCl}(\text{aq}) + \text{NaOH}(\text{aq}) \rightarrow \text{H}_2\text{O}(\text{l}) + \text{NaCl}(\text{aq})$
 $\text{FeBr}_3(\text{aq}) + \text{H}_3\text{PO}_4(\text{aq}) \rightarrow \text{FePO}_4(\text{s}) + 3\text{KBr}(\text{aq})$

The chemical equations in

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Model 1 contain the phase notations G), 0), G), and (aq). Match.

Chemical - LTHS Answers

reactions that occur daily in living things and in laboratories. Knowing the main categories of chemical reactions can help you to remember and understand them. It also helps you to recognize

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pattern and predict the products of many chemical reactions. Model 1: Types of Chemical Reactions Type of Reaction Symbols Chemical Equations

POGIL “High School Drama” Model 1: Types of Chemical Reactions

The four sets of chemical reactions shown in Model 1 have the following

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general names. Discuss within your group which name belongs to which set of chemical reactions. Write the name in the appropriate place in Model 1. Single Replacement Reaction Synthesis Reaction Double Replacement Reaction Decomposition Reaction 7.

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INFORMATION There are 5 general types

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of chemical reactions. Those types are synthesis (or direct combination), decomposition, single replacement, double replacement (or metathesis), and combustion. Each of the five reaction types has a unique template that helps to set it apart from the other four.

POGIL 03 - Chemical Reactions 3 -

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Types of Reactions

- Identify and differentiate between four types of chemical reactions: synthesis, decomposition, single replacement and double replacement.

Types of Chemical Reactions

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Answers

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The answer comes in a type of reaction called the oxidation reduction or redox reaction. Oxidation and reduction pogil answers pogil general chemistry at

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franklin marshall college. This is a reaction in which electrons are transferred between reactants. Page 1/6

Oxidation And Reduction Pogil Answers

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Assigned as CW on 2/7/20 . Formulas and Equations Review Distributed on 2/10/20.

Piersa, Amanda / Unit 7: Formulas, Equations and Chemical ...

17. Consider the following chemical reaction written as a word equation.
diphosphorus pentoxide + water →

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phosphoric acid. a. Identify the type of chemical reaction from Model 1 that would describe this reaction. Synthesis.
b. Write chemical formulas under the names of the substances in the word equation. $P_2O_5 + 3H_2O \rightarrow H_3PO_4$

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The four sets of chemical reactions shown in Model 1 have the following general names. Discuss within your group which name belongs to which set of chemical reactions. Write the name in

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the appropriate place in Model 1. Single Replacement Reaction Double Replacement Reaction Synthesis Reaction Decomposition Reaction

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Chemical reactions can be classified into different categories. Four common types

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are synthesis, decomposition, single replacement and double replacement. Specific reactions corresponding to these general types are associated with health issues, environmental problems, and manufacturing process.

Shall We Dance? Classifying Types of Chemical Reactions

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Chemical Reactions Question Type Essay
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Using a POGIL activity, students will explore four types of reactions:

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synthesis, decomposition, single displacement, and double displacement.

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